SYNTHESIS OF SOME TRIFLUOROMETHYL PYRIDO-THIAZOLIDINONES AND THEIR ANTIFUNGAL ACTIVITY

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ABSTRACT

2-Aryl-3-(2-substituted amino 3, 5-disubstituted pyridin)-1,3-thiazolidin-4-ones (2) have been synthesized by crystallization of Arylaldehyde-2- (3,5 disubstituted) pyridin-2-yl hydrazones (1) with mercapto acetic acid in dioxane. The compounds have been screened for their fungicidal activities against A. niger and A. flavus.

Keywords : Thiazolidinone, pyridin, trifluoro methyl group.

The chemistry and diverse biological activity¹ exhibited by thiazolidinone derivatives have been reviewed. Reports having the fungicidal, antimicrobial and other biological activities of Thiazolidinones are also available. Further trifluoromethyl group and pyridine incorporating toxophores and are expected to display strong fungicidal activities.

With this objective in mind and in continuation of our work on the synthesis of biologically active heterocyclic compounds²⁻⁴, the synthesis of above compounds were undertaken.

EXPERIMENTAL

All melting points are uncorrected. I R spectra were recorded on a Perkin-Elmer 157 spectrophotometer and PMR spectra on Perkin-Elmer R-32 spectrometer at 60 MHz.

1. <u>Arylaldehyde-2 (3, 5-disubstituted) pyriden-2-yl-1-substituted hydrazone (I b)</u>

A mixture of 1-(3. trifluoromethyl) pyriden-2-yl hydrazine (0.01 M) and 4methoxylbenzaldehyde (0.01 M) was reflexed in methanol for an hour in presence of gl. Acetic acid. It was cooled and poured into water. The product thus separated out was filtered and recrystallized from ethanol. mp (72^{0}) IR (KBr) cm⁻¹ 1620 (C=N), 1590, 1570, 1550 (aromatic ring). PMR (DMSO-d₆) δ : 2.1 (s, 3H, CH₃), 2.4 (s, 3H, -OCH₃), 4,5 (s 1H, -CH), 7.2-8.0 (m, 6H-Ar-H).

Other compounds were prepared similarly (Scheme-1).

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2. <u>2-Aryl-3-(2-substituted amino 3, 5-disubstituted pyriden)-(1,3-thiazolidin-4-ones</u> <u>II b)</u>

A mixture of I (.01 M) and mercaptoacetic acid (.01M) was refluxed in 1,4-Dioxane for 3 hours. The solvent was removed and residue was poured in water and neutralized by NaHCO₃. The solid mass was then recrystallized by ethanol. m.p (116^o C), IR (KBr) cm⁻¹ 1640 (>C =0), 1580, 1570, 1550 (aromatic ring). PMR (DMSO) δ 2.1(s, 3H, CH₃), 2.5 (s, 3H, -OCH₃), 3.5 (s, 2H, -CH₂CO), 3.1(s, 1H, s- CH-N), 7.2 -8.5 (m, 6H, Ar-H).

Other compounds were prepared similarly (Scheme-1)



(II)

<u>Scheme 1</u>

a: $R_1 = H$, $R_2 = CF_3$, $R_3 = CH_3$, $R_4 = 4-CH_3$ **b:** $R_1 = H$, $R_2 = CF_3$, $R_3 = CH_3$, $R_4 = 4-OCH_3$ **c:** $R_1 = CF_3$, $R_2 = CI$, $R_3 = H$, $R_4 = 4-CH_3$ **d:** $R_1 = CF_3$, $R_2 = H$, $R_3 = H$, $R_4 = 4-CI$ **e:** $R_1 = CI$, $R_2 = CF_3$, $R_3 = CH_3$, $R_4 = 4-OCH_3$ (IJISE) 2015, Vol. No. 2, Jul-Dec

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Compound	Percentage Inhibition after 96 hours					
	Organis	A. Niger		A . flavus	Used	
	m	Used		cone		
	Cone					
	500	100	10	500	100	10
	ppm	ppm	ppm	ppm	ppm	ppm
Ia	75	62	50	78	60	52
I _b	60	50	42	62	55	45
II _d	90	80	66	88	79	62
IIc	82	70	60	83	72	60
IIb	78	68	50	76	66	55
carbendazim	96	85	73	95	82	70

Table 1: Fungicidal activity of compound I & II

FUNGICIDAL ACTIVITY

The fungicidal activity of I a-d and II a-d was evaluated by agar growth technique⁵ of 500, 100 and 10 ppm concentration against the fungi A. niger and A. flavus using carbendazim a commercial fungicide, as standard. The results are recorded in Table 1.

All the compounds exhibited moderate fungi toxicity which decreased on dilution. Compound no. II d showed the highest activity (90% at 500 ppm and 66% at 10 ppm) which was quite comparable to the commercial carbendazim (96% at 500 ppm and 73 % at 10 ppm) tested under similar conditions. Further investigation of this compound on a wider range and at more dilution is desirable.

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